

<b>Acid/Base Neutralization</b>	<b>Identifier:</b>	<b>CSP-0001</b>
	<b>Revision:</b>	<b>1</b>
	<b>Page:</b>	<b>1 of 2</b>

ISU Chemistry Department	Stockroom Procedure	Effective Date: 06/16/2020
--------------------------	---------------------	----------------------------

## 1. INTRODUCTION

This procedure guides in the neutralization of acids and bases in the ISU Chemistry Stockroom.

## 2. PRECAUTIONS AND LIMITATIONS

- 2.1. Add acid or base to water or a dilute bicarbonate solution.
- 2.2. Wear proper personal protective equipment (PPE).

## 3. APPARATUS AND MATERIALS

- 3.1. 4L or larger plastic container
- 3.2. Stir bar retriever
- 3.3. pH paper
- 3.4. Gloves
  - 3.4.1. Nitrile for common acids and bases
  - 3.4.2. Heavy/Butyl rubber gloves for sulfuric acid or NoChromix solution

## 4. REAGENTS

- 4.1. Baking soda

## 5. INSTRUCTIONS

- 5.1. Add approximately 500 mL of water to a 4L container.
- 5.2. Add 1/3 Cup of baking soda to the water in the 4L container.
  - 5.2.1. Mix to dissolve.
- 5.3. Slowly add acid/base to the baking soda solution and stir.
- 5.4. For **Acids**:
  - 5.4.1. When solution stops fizzing, check pH with pH paper
    - 5.4.1.1. If pH is between 4 and 10:
      - 5.4.1.1.1. The solution may be poured down the drain.
    - 5.4.1.2. If pH is not between 4 and 10:
      - 5.4.1.2.1. Add more baking soda and stir.
      - 5.4.1.2.2. Return to step 5.4.1.



<b>Acid/Base Neutralization</b>	<b>Identifier:</b>	<b>CSP-0001</b>
	<b>Revision:</b>	<b>1</b>
	<b>Page:</b>	<b>2 of 2</b>

ISU Chemistry Department	Stockroom Procedure	Effective Date: 06/16/2020
--------------------------	---------------------	----------------------------

**NOTE:** For high concentrations of acids, a dilute base may be added slowly in 10-20 mL increments. Use caution as the solution will get hot & may flash boil. Perform work in a fume hood.

5.5. For **Bases:**

- 5.5.1. Add approximately 1 L of the base to be neutralized into the baking soda solution from step 5.2.
- 5.5.2. Add any "Used acids" available to further neutralize the solution down to pH 10 or under.
- 5.5.3. If pH is less than 10:
  - 5.5.3.1. The solution may be poured down the drain.
- 5.5.4. If pH is more than 10:
  - 5.5.4.1. Add more acid and recheck pH
    - 5.5.4.1.1. Return to step 5.5.2

**Note:** If no "Used acids" are available, use 1M HCl.

**6. Cleanup**

- 6.1. Rinse "Used Acid/Used Base" bottles with 5% sodium bicarbonate solution followed by deionized water.
- 6.2. Return the labeled "Used Acid/Used Base" bottles to storage area.